



Basic Electricity – Unit 3 Part 1: Atomic Structure

Worksheet 1

How many electrons fit in each shell around an atom?

The maximum number of electrons that can occupy a specific energy level can be found using the following formula:

$$\text{Number of electrons in a shell} = 2n^2$$

The variable n represents the Principal Quantum Number, the number of the energy level in question.

Energy Level (Principal Quantum Number)	Shell Letter	Electron Capacity
1	K	2
2	L	8
3	M	18
4	N	32
5	O	50
6	P	72





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