EICC COURSE DEVELOPMENT MODEL (CDM)

CATALOG COURSE NUMBER: MFG-190 COURSE TITLE: Metallurgy Originating College: CCC GMCC SCC Initiating Faculty Member: Kenneth Darmody

Effective Term/Year: Fall 2014 Initiating Department Coordinator: Ben Kettering

Reason for submission: Check all that apply

New Course If yes, type of course:
A&S

To be considered for General Education?
Yes INO Category:
To be part of an A & S Concentration?
Yes INO Concentration:
CTE Program Title: IRequired Elective

General Education or Program Review IReactivation of an inactive course IMaking course inactive
Changing course; please explain:
Other; please explain:

Contact Hours/Distribution of Contact Hours

Lecture Hours Lab Hours **Clinical Hours Coop Hours** Hours per Week: 1.00 Hours per Week: 2.00 Hours per Week: 0 Hours per Week: 0 Number of Weeks: 16.50 Number of Weeks: 16.50 Number of Weeks: 16.50 Number of Weeks: 16.50 **Note: If offering a course for the full fall or spring semester, the number of weeks is 16.5 Total Lecture Hrs: 19.80 Total Lab Hrs: 39.60 Total Clinical Hrs: 0 Total Coop Hrs: 0

Semester Hours Credit: 2.00 if variable credit, give range:

Allow repeat^{*} for credit: □Yes ☑No

If yes, total course repeats allowed: If yes, total credits:

*Note that repeat for credit means a student can pass the course and then repeat it for additional credit. An internship course is an example of a course that could be set up as repeatable for additional credit

Course or courses this CDM replaces, if any:

CATALOG COURSE DESCRIPTION: This course teaches students the basic theory of ferrous and non-ferrous metals. In addition, this course focuses on how metals differ in terms of hardness, brittleness, durability, resistance to corrosion, machinability and weldability.

RECOMMENDED ENTRY LEVEL SKILLS/KNOWLEDGE:

PRE-REQUISITE COURSES

CCN#	COURSE TITLE		
CO-REQUISITE COURSES			

CCN#	COURSE TITLE

PUBLISHED MATERIAL(S) USED FOR CDM DEVELOPMENT: Kibbe, Richard, John Neely, Warren White, and Roland Meyer. Machine Tool Practices. Upper Saddle River: Prentice Hall, 2010. Series 100 Fundamentals-Course 105:Metals in the Plant. Buffalo Grove: TPC Training Systems, 2012.

In general it is expected that source material will be dated within 5 years of this CDM date. If all materials/ textbooks cited above are older than this, please explain:

GENERAL COURSE GOALS

Upon successful completion of this course the student should be able to:

Identify ferrous metals. Identify non-ferrous metals. Identify properties of metals. Identify types of iron and steel and their properties. Identify types of steel and their individual properties. Understand heat treating processes. Identify various grades of copper, aluminum, magnesium and titanium and their various properties. Identify various grades of lead, nickel, tin and zinc and their individual properties.

TOPICAL OUTLINE

- 1. Introduction to Metals
- 2. Properties of Metals
- 3. Manufacturing Processes
- 4. Properties of Iron and Steel
- 5. Properties of Standard Steels
- 6. Heat Treating
- 7. Properties of Copper
- 8. Properties of Aluminum
- 9. Properties of Magnesium and Titanium
- 10. Properties of Lead, Nickel, Tin and Zinc

COURSE OBJECTIVES

Upon successful completion of the course, a student should be able to:

- 1. Introduction to Metals
 - a. Identify five metals or alloys commonly used in industry.
 - b. Identify five mechanical properties of metals.
 - c. Describe the uses of three metal alloys.
- d. Describe the metal working processes of casting, forming and machining.
- 2. Properties of Metals
 - a. Identify the definitions of four mechanical properties of metals.
 - b. Describe the three kinds of stress.
 - c. Identify the ways in which a metal can fail.
- 3. Manufacturing Processes
 - a. Identify four kinds of molds used in casting.
 - b. Describe extrusion.
 - c. Identify the steps involved in making a part by powder metallurgy.
 - d. Describe the differences between hot-chamber and cold-chamber die-casting.
- 4. Properties of Iron and Steel
 - a. Identify the commercial grades of cast iron.
 - b. Identify the mechanical properties of the grades of cast iron.
 - c. Describe the forms in which carbon appears in commercial grades of cast iron.
 - d. Describe the process of smelting.
- 5. Properties of Standard Steels
 - a. Describe the definition of steel.
 - b. Describe how to conduct a spark test.
 - c. Identify steel sheets having as-rolled edges and cut edges.
- d. Describe two differences between alloy steels and steels containing only iron and carbon.
- 6. Heat Treating
 - a. Describe the two basic processes of heat treating.
 - b. State the four major purposes of heat treating.
 - c. Identify why distortion and cracking occur during welding.
 - d. Define low, medium and high carbon steels.
 - e. Describe how to anneal, harden and temper a star drill.
- 7. Properties of Copper
 - a. Identify the steps in producing copper from ore.
 - b. Describe the contents of red brass and its uses.

c. Identify the three groups of brasses, based on their zinc content and the three categories of hardness.

8. Properties of Aluminum

- a. Define wrought-grade and casting grade aluminums.
- b. Identify the advantages of aluminum-silicon alloys.
- c. Describe how aluminum is anodized.
- d. Identify the advantages and disadvantages of the oxide coating on aluminum.
- 9. Properties of Magnesium and Titanium
 - a. Identify the alloys of magnesium.
 - b. Identify the useful properties of magnesium and titanium.
 - c. Identify the alloys of titanium.
 - d. Describe how to join magnesium alloys.
- e. Describe the precautions that must be taken when working with magnesium and titanium.
- 10. Properties of Lead, Nickel, Tin and Zinc
 - a. Describe the properties and characteristics of lead.
 - b. Describe how tinplate is manufactured.
 - c. Describe how zinc is refined and processed.
 - d. Identify the properties that are improved by adding nickel to stainless steel.

RECOMMENDED METHODS OF INSTRUCTION: Check all appropriate methods of instruction to facilitate student learning of course objectives.

□Case Studies	Class Discussions			
Computer lab work	Computer-assisted tools			
Computer-assisted writing	Conducting experiments			
Demonstration or modeling	Electronic interaction			
Field observation	□Field trips			
□Guest speaker	Guided practice			
□In-class writing or editing workshops	□Journals			
■Lecture	Library instruction and resources			
Model building	□Peer review			
□Readings	□Role play			
□Service learning	□Simulation			
Student and instructor conferences	Student collaborative learning			
Student presentation	Student projects			
Tests or quizzes	□Worksheets/surveys			
Uriting assignments/exercises (graded or not)				
Other (please list specifics): Videotapes, Mill Projects				

RECOMMENDED EVALUATION METHODS: Check all appropriate methods of evaluation to assess student achievement of course objectives.

Class workshops	Classroom discussions/participation		
Collaborative work	Demonstration of skill(s)		
Individual conferences	□Journals		
Laboratory reports	Oral presentations		
□Portfolios	□Pretest/Posttest		
Quizzes	Reading responses		
Student presentations	Student projects		
☑Tests	Writing Assignments		

□Other (please list specifics):

ATTENDANCE: Policies on attendance will be formulated by the instructor and communicated to the students on the course syllabus.

Date:

ACADEMIC DISHONESTY: Policies on academic dishonesty can be found in the EICC student code of conduct published in the student handbook.

Originally Written by:

Department Chair, Comments, & Date:

Does similar curriculum exist at other EICC Colleges?
CCC
MCC
SCC
No

If yes, Counterparts Consulted, College, Comments & Date:

CDM Review or Revision Date:

Faculty member(s) & College:

Does similar curriculum exist at oth	er EICC Colleges? □CCC	□MCC	□SCC	⊡No
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Changes made to course which will require further review steps:

Making course inactive
Credit hours
Contact hours
Course Description

□ 25% or more of course objectives □ Other minor revisions or no revisions

Dean Review, Comments & Date:

If changes made require further review and approval:

College Curriculum Committee Sign-off & Date:

IC Review Subcommittee Sign-off & Date:

Instructional Council Approval: